

Eine theoretisch interessante Möglichkeit zur Beeinflussung der Gemischtrennung bietet auch die in Gl. (19) und (20) wegen der Annahme $T = \text{const}$ nicht berücksichtigte Thermodiffusion. Wegen des in weiten Bereichen der Trenndüse von der Umlenk wand nach innen gerichteten Temperaturgefälles¹³ dürfte sie im Prinzip zu einer Verminderung der UF_6 -Sedimentation führen. Da eine entsprechende, gegen den Zentrifugaleffekt gerichtete Thermodiffusion der Isotope wahrscheinlich zu vernachlässigen

ist¹⁴, könnte die Thermodiffusion über das leichte Zusatzgas auf indirektem Wege zur Entmischung der Uranisotope beitragen.

Abschließend sei noch darauf hingewiesen, daß in der Trenndüse wegen der Kombination von großer Mach-Zahl und großer Knudsen-Zahl ein merklicher Einfluß gaskinetischer Effekte höherer Ordnung zu erwarten ist. Rechnungen nach dem Momentenverfahren haben gezeigt, daß sich daraus meßbare Beiträge zur Isotopentrennung ergeben können.

¹³ U. EHRFELD, Interner Bericht des Instituts für Kernverfahrenstechnik KVT 55/25 [1970].

¹⁴ Nach Messungen von P. KIRCH und R. SCHÜTTE (Z. Naturforsch. **22a**, 1532 [1967]) liegt der Thermodiffusionsfaktor von gasförmigem UF_6 nur in der Größenordnung von

einigen 10^{-5} . Nach Untersuchungen von S. K. DEB und A. K. BARUA (Phys. Fluids **10**, 992 [1967]) ist mit keiner wesentlichen Veränderung des Thermodiffusionsfaktors durch das leichte Zusatzgas zu rechnen.

⁷³Ge Nuclear Magnetic Resonance Studies

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(Z. Naturforsch. **26a**, 1384—1389 [1971]; received 12 June 1971)

The ⁷³Ge spectra of some germanium-tetra-halides, germanium-tetra-alkyles, and of germanium-tetra-methoxide have been observed. The chemical shifts of these ⁷³Ge NMR-lines were measured as functions of the temperature in the range up to 50 °C. In CS_2 - and benzene-solutions of GeI_4 , two ⁷³Ge-lines were found. The ⁷³Ge NMR-spectrum of $\text{Ge}(\text{CH}_3)_4$ caused by indirect spin-spin-coupling between the ⁷³Ge nucleus and the protons is well resolved. The spin-spin-coupling constants of $\text{Ge}(\text{CH}_3)_4$ and of $\text{Ge}(\text{OCH}_3)_4$, which shows an unresolved spectrum, were evaluated by an iterative least square fitting routine. The relaxation times T_2 (in the range 30...740 msec) measured by the Carr-Purcell spin-echo-technique and calculated from the line widths are in good agreement. Precision measurements of the ratios of the Larmor frequencies of ⁷³Ge in GeCl_4 , ⁴¹K in aqueous KF-solution and ²H in pure heavy water lead to a magnetic moment of ⁷³Ge: $\mu = -(0.87678 \pm 0.00001) \mu_N$ (uncorrected).

1. Introduction

The ⁷³Ge nucleus (natural abundance 7.6%, and Larmor frequency 2.7 MHz in a field $B_0 \approx 18$ kG) delivers a weak NMR-signal; it is about 6 orders of magnitude weaker than the NMR-signal of protons in H_2O in the same magnetic field B_0 , and with the same probe volume. There are only three ⁷³Ge NMR-studies¹⁻³ on GeCl_4 , but no high resolution ones on other compounds.

⁷³Ge has a nuclear spin $I = 9/2$ and an electrical quadrupole moment $Q = -0.29$ barn⁵. Therefore

one expects broad resonance lines of molecules with unspherical symmetry, i. e. a rapid decay of the NMR-signals; e. g. the NMR-signal of methyl-germanium-trichloride decays so rapidly that we were not able to find it. This yields an upper limit for the relaxation time $T_2 < 100 \mu\text{sec}$.

On the other hand, narrow NMR-lines are expected from germanium compounds with spherical symmetry. As there exist some NMR-investigations of tetrahedral compounds of the Group IV-elements e. g. ¹³C, ²⁹Si, and ¹¹⁹Sn, we investigated such germanium compounds.

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¹ C. D. JEFFRIES, Phys. Rev. **92**, 1262 [1953].

² S. I. AKSENOV and K. V. VLADIMIRSKII, Dokl. Akad. Nauk SSSR **96**, 37 [1954].

³ O. LUTZ, A. SCHWENK, and G. ZIMMERMANN, Phys. Letters **25A**, 653 [1967].

⁴ J. M. MAYS and C. H. TOWNES, Phys. Rev. **81**, 940 [1951].

⁵ W. J. CHILDS and L. S. GOODMAN, Phys. Rev. **141**, 15 [1966].



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Although there would be some interest to investigate mixed Ge-halides, as this was done by BURKE and LAUTERBUR⁶ with tin(IV)-halides, it is not worth while with the ^{73}Ge resonance, as the line widths of unsymmetrical compounds are larger than the chemical shifts of these lines in usual fields B_0 .

2. Experimental

2.1. Apparatus

The apparatus, developed to detect very weak NMR-signals, is a pulsed spectrometer described in ^{7,8}. The magnetic field $B_0 = 18.07$ kG is stabilized with a ^7Li -NMR-probe⁹. The spatial homogeneity is achieved by special nickel-shims. The probe assembly, with a one-coil-arrangement, permits a sample rotation up to 270 Hz. By a thermostat the temperature of the sample is to be held constant within an accuracy of $\pm 1^\circ\text{C}$.

A resolution of $2 \cdot 10^{-7}$ is obtained with this probe assembly and with a sample volume of 1.5 cm^3 , i. e. at the ^{73}Ge resonance frequency of 2.7 MHz the width of a NMR-line caused by the inhomogeneity is less than 0.4 Hz, on the other hand, relaxation times $T_2 \approx 200$ msec can be determined from the free-precession decay of the NMR-signal with an influence less than 10% due to the inhomogeneity of the magnetic field.

The measurements were done in such a manner that the NMR-signal decays completely during the time interval (0.1...2 sec) between the rf-pulses. The signal/noise-ratio was improved by using a time-averaging-computer (Signal Analyzer 5480 A of Hewlett Packard).

2.2. Evaluation of the NMR-signals

The signal stored in the time-averaging-computer is transferred to a computer (CDC 3300 of the ZDV, Tübingen) by punched tape to calculate the Fourier-Transform of the averaged data. The real part of this Fourier-Transform, the absorption spectrum is a Lorentzian line with a half width:

$$\Delta\nu_{1/2} = 1/(\pi T_2) \quad (1)$$

in the case of an exponentially decaying free-precession-signal, T_2 is the transverse relaxation time.

Shifts were measured by the sample replacement method. The evaluation of the Carr-Purcell spin-echos was done by Fourier-Transformation of each echo, i. e. of the NMR-signal between two successive 180° -pulses respectively. Time intervals between the 180° -pulses were chosen in such a way, that the decay of the NMR-signal due to the magnetic field inhomogeneity was small. The maximum of the absolute value of the Fourier-Transform of an echo is proportional to the height of this echo. From the decay of these maxima T_2 is calculated.

The determination of absolute values of spin-spin-coupling constants was done by an iterative least square fitting routine. The number of lines in a spectrum and their relative intensities are well known; under this assumption, the following parameters of the spectrum can be calculated by this routine: the centre of the line group, the coupling constant, the width of a single line, and the absolute intensity of the lines.

2.3. Samples

The samples were all contained in cylindrical glass tubes of 9 mm internal diameter, and most were measured at a temperature of $(30 \pm 1)^\circ\text{C}$. At this temperature all of them, with the exception of GeI_4 were liquids. GeBr_4 which melts at 26°C was undercooled for measurements at deeper temperatures.

The GeI_4 samples were a 2.3-molar solution in CS_2 and a 0.57-molar solution in benzene. These concentrations correspond nearly with the saturated solutions at a temperature of 40°C which was the measuring temperature of the two GeI_4 samples.

GeCl_4 and tetra-methyl-germane (GeMe_4) were delivered by the Schuchardt GmbH, München. GeBr_4 , GeI_4 , tetra-ethyl-germane (GeEt_4), tetra-propyl-germane (GePro_4), tetra-butyl-germane (GeBu_4), and $\text{Ge}(\text{OCH}_3)_4$ were manufactured by the Alfa Inorganic, Inc., Beverly. The purities of the germanium-halides are 99.999%, those of the organometallic compounds are in the range 97...99.9%.

3. Results

3.1. The Ratio of the Larmor Frequencies of ^{73}Ge , ^{41}K , and ^2H

Our former measurements of the Larmor frequency³ of ^{73}Ge were carried out with a pulse period T , small compared with the relaxation time T_2 . This pulse technique, described in ⁸ as "Fall 3", avoids signal loss due to the inhomogeneity of the field B_0 , the NMR-lines are broadened by saturation, but in an inhomogeneous field B_0 shifts between the maximum of such a line and the Larmor frequency may occur¹⁰. This effect was underestimated at that time. Now we were able to repeat those measurements with a smaller and rotating probe sample, the pulse period $T \approx 1$ sec was now longer than the relaxation times T_1 and T_2 , therefore such errors are eliminated.

The Larmor frequencies of ^{73}Ge , and ^{41}K were measured with the pulse spectrometer. The Larmor

⁶ J. J. BURKE and P. C. LAUTERBUR, J. Amer. Chem. Soc. **83**, 326 [1961].

⁷ J. KAUFMANN and A. SCHWENK, Phys. Letters **24 A**, 115 [1967].

⁸ A. SCHWENK, Z. Phys. **213**, 482 [1968].

⁹ J. KAUFMANN and A. SCHWENK, Z. Angew. Phys. **21**, 527 [1966].

¹⁰ A. SCHWENK, J. Magn. Resonance, in press.

frequency of ^2H was detected with a normal crossed-coil NMR-spectrometer*. The samples were GeCl_4 , a 9-molar aqueous solution of KF and pure heavy water.

From about 40 measurements with each probe, the following results, corrected for spherical probe shape, are obtained:

$$\nu_{\text{Ge}}/\nu_{\text{K}} = 1.361\,966\,4 \pm 0.000\,000\,5.$$

$$\nu_{\text{Ge}}/\nu_{\text{H}} = 0.227\,248\,6 \pm 0.000\,001\,0,$$

$$\nu_{\text{K}}/\nu_{\text{H}} = 0.166\,853\,1 \pm 0.000\,000\,8.$$

The errors of the second and the third result are the sum of the following errors: three times the r.m.s. error, an error of ± 2 ppm, caused by the different field distributions across the two samples and an error of ± 2 ppm, which arises from the uncertainty in the magnetic field value within the two different probe assemblies. The error of the first result is three times the r.m.s. error, as both Larmor frequencies were determined with the same probe assembly. With the third and the first results one gets

$$\nu_{\text{Ge}}/\nu_{\text{H}} = 0.227\,248\,3 \pm 0.000\,001\,0$$

which is in good agreement with the value of the second result.

Using the nuclear spin of ^{73}Ge , $I = 9/2$, the ratio $\nu_{\text{H}}/\nu_{\text{H}}$ of SMALLER¹¹, and the proton moment (uncorrected for H_2O) of COHEN and DuMOND¹², and the sign detected by JEFFRIES¹, one gets the magnetic moment of ^{73}Ge in GeCl_4 without corrections:

$$\mu_{\text{Ge}} = - (0.876\,78 \pm 0.000\,01) \mu_{\text{N}}.$$

3.2. Chemical Shifts

The ^{73}Ge chemical shifts were measured in the tetra-halides and some organometallic compounds. The results are given in Table 1. The chemical shift is defined:

$$\delta_x = (\nu_{\text{ref}} - \nu_x) / \nu_{\text{ref}}$$

(both frequencies measured at the same field B_0).

All of the shifts are reported as parts per million (ppm), relative to the resonance frequency of the reference GeCl_4 . The values are corrected to spherical probe-shape, except that of $\text{Ge}(\text{OCH}_3)_4$, the corrections are less than 0.5 ppm. The error is three times the r.m.s. error.

In both solutions of GeI_4 , there are two lines of the ^{73}Ge resonance with the intensity ratio $\approx 4:1$. The stronger line is due to the molecule GeI_4 , whereas the weaker line could be due to a chain compound of the type $\text{I}_3\text{Ge}-\text{GeI}_3$ or to GeI_2 . This assumption is based on the fact that free iodine was pointed out in both solutions.

Table 1. ^{73}Ge chemical shifts of germanium-tetra-compounds.

Compound	Solvent	Chemical shift/ppm	Remarks
GeCl_4	—	0	reference
GeBr_4	—	343.0 ± 0.3	—
GeI_4	benzene	1139.0 ± 0.5	stronger line
	benzene	852 ± 3	weaker line
GeI_4	CS_2	1111.6 ± 0.5	stronger line
	CS_2	831 ± 5	weaker line
$\text{Ge}(\text{OCH}_3)_4$	—	66.9 ± 0.2	—
GeMe_4	—	30.9 ± 0.5	—
GeEt_4	—	12.8 ± 0.2	—
GePro_4	—	28.8 ± 0.3	—
GeBu_4	—	25.4 ± 0.3	—

3.3. Temperature Dependence of Chemical Shifts

All compounds mentioned above, except GeMe_4 , and GeI_4 , were investigated for a temperature dependence of the chemical shift of the ^{73}Ge resonance. Within the limits of error, neither the chemical shift of all germanium-alkyles in the range $20 \dots 40^\circ\text{C}$, nor the shift of GeCl_4 in the range $0 \dots 50^\circ\text{C}$ depend on the temperature. A dependence of the chemical shift on the temperature Θ was found at GeBr_4 and $\text{Ge}(\text{OCH}_3)_4$. The shifts are to be described by the following functions:

$$\delta(\Theta) = 341.1 \text{ ppm} + (0.059 \text{ ppm}/^\circ\text{C}) \cdot \Theta$$

for GeBr_4 in the range $20^\circ\text{C} \leq \Theta \leq 40^\circ\text{C}$

$$\text{and } \delta(\Theta) = 66.2 \text{ ppm} + (0.026 \text{ ppm}/^\circ\text{C}) \cdot \Theta$$

for $\text{Ge}(\text{OCH}_3)_4$ in the range $0^\circ\text{C} \leq \Theta \leq 50^\circ\text{C}$. The error is maximum ± 0.2 ppm.

3.4. Comparison of the Chemical Shifts of the Group IV-Elements ^{13}C , ^{29}Si , ^{73}Ge , and ^{119}Sn

^{13}C , ^{29}Si , and ^{119}Sn -chemical shifts of the alkyles and halides were measured by HUNTER and REE-

* We like to thank Dr. O. LUTZ for performing the ^2H measurements.

¹¹ B. SMALLER, Phys. Rev. **83**, 812 [1951].

¹² E. R. COHEN and J. W. DuMOND, Rev. Mod. Phys. **37**, 537 [1965].

VES¹³, LAUTERBUR^{14, 15}, and BURKE and LAUTERBUR⁶. Figure 1 shows the shifts of the alkyls as measured till now. In Fig. 2, the shifts of the halides are plotted. The shifts of all elements in both figures, are related to the tetra-chloride as reference. The chemical shifts between the various germanium-

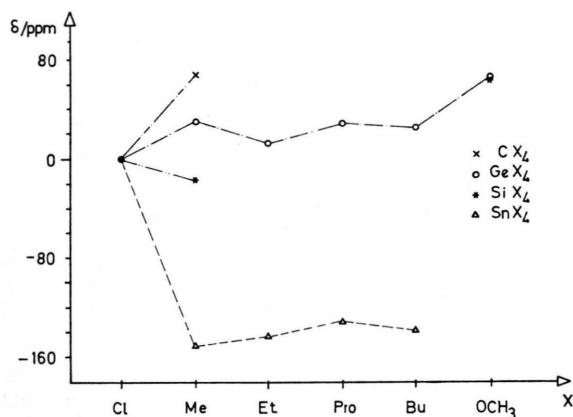


Fig. 1. Chemical shifts of some tetra-alkyls of the Group IV-elements ^{29}Si , ^{73}Ge , and ^{119}Sn related to the tetra-chloride. The ^{29}Si -measurements were performed by HUNTER and REEVES¹³, and those of ^{119}Sn by BURKE and LAUTERBUR⁶, and HUNTER and REEVES¹³.

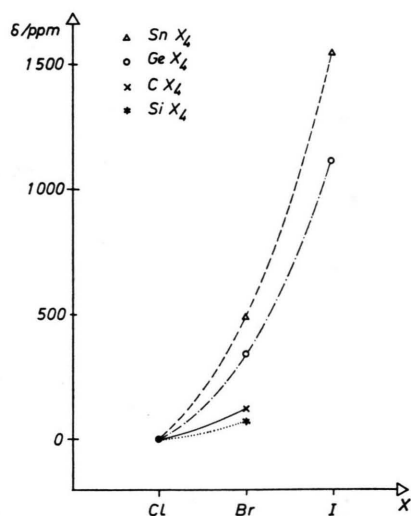


Fig. 2. Chemical shifts of the tetra-halides of the Group IV-elements ^{13}C , ^{29}Si , ^{73}Ge , and ^{119}Sn related to the tetra-chloride. The ^{13}C -measurements were performed by LAUTERBUR¹⁴, those of ^{29}Si by LAUTERBUR¹⁵ and HUNTER and REEVES¹³, and those of ^{119}Sn by BURKE and LAUTERBUR⁶.

alkyls are relatively small just as the shifts of the other alkyls yet measured. In relation to the GeCl_4 , the germanium alkyls show a slightly positive shift, whereas the resonances of the tin-alkyls are more than 130 ppm negatively shifted in relation to the SnCl_4 resonance. The shifts of the tetra-halides of the elements mentioned above show a similar trend; e. g. between the chemical shifts of the tin-halides δ_{SnX_4} and those of the germanium-halides, the following relation holds:

$$\delta_{\text{SnX}_4} = 1.41 \delta_{\text{GeX}_4} \quad (\text{X} = \text{Br}, \text{I}).$$

3.5. Indirect Spin-Spin-Coupling Constants

The protons in the alkyl-groups of the organo-metallic Ge-compounds are coupled to the ^{73}Ge nucleus by indirect spin-spin-coupling. Figure 3 shows the ^{73}Ge NMR-spectrum of $\text{Ge}(\text{CH}_3)_4$.

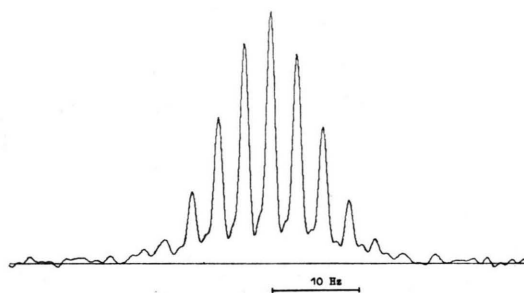


Fig. 3. ^{73}Ge NMR-spectrum of $\text{Ge}(\text{CH}_3)_4$.

As all protons are equivalent spins, this spectrum consists of 13 lines with the intensities proportional to the binomial coefficients $\binom{12}{k}$, ($k=0, \dots, 12$). 9 of these lines are to be seen in this spectrum, the other ones are less than the noise. From 12 such spectra, the absolute value of the coupling constant was evaluated by the fitting routine:

$$J_{^{73}\text{Ge}-^1\text{H}} = (2.99 \pm 0.03) \text{ Hz for } \text{Ge}(\text{CH}_3)_4.$$

The error is the maximum error.

This coupling constant has also been determined in 1963 by SMITH¹⁶, and in 1964 by SZALMONA¹⁷ from the proton NMR-spectrum of GeMe_4 . The results are:

$$\begin{aligned} J &= (2.92 \pm 0.02) \text{ Hz} && (\text{Smith}), \\ J &= 2.94 \text{ Hz} && (\text{Szalmona}). \end{aligned}$$

¹³ B. K. HUNTER and L. W. REEVES, Can. J. Chem. **46**, 1399 [1968].

¹⁴ P. C. LAUTERBUR, J. Chem. Phys. **26**, 217 [1957].

¹⁵ P. C. LAUTERBUR, in: Determination of Organic Structures by Physical Methods, edit. by F. C. NACHOD and W. D. PHILLIPS, Academic Press, New York 1962, Vol. 2.

¹⁶ G. W. SMITH, J. Chem. Phys. **39**, 2031 [1963].

¹⁷ A. SZALMONA, Molec. Phys. **7**, 497 [1964].

A similar case as described here is found at germanium-tetra-methoxide $\text{Ge}(\text{OCH}_3)_4$; this compound should also deliver a NMR-spectrum of 13 lines. The absorption curve of $\text{Ge}(\text{OCH}_3)_4$ is essentially different from a Lorentzian shape as recorded at all Ge-halides; this leads to the supposition of an unresolved spectrum. Indeed, the least square fitting routine delivers besides the absolute value of the coupling constant of $\text{Ge}(\text{OCH}_3)_4$

$$J_{\text{Ge}-^1\text{H}} = (1.9 \pm 0.3) \text{ Hz}$$

the half width of a single line $\Delta\nu_{1/2} = (12.2 \pm 1.5) \text{ Hz}$, which is in good agreement with the relaxation time $T_2 = (30 \pm 3) \text{ msec}$ measured by the Carr-Purcell method (see Sect. 3.7).

The other germanium alkyls investigated here, have not only one kind of equivalent protons; a more complicated structure of the NMR-spectrum results from this. The recorded absorption-curves show only one NMR-line, which differs strongly from a Lorentzian shape i. e. there is an unresolved spectrum; the half width of the NMR-line is 3...7 times greater than the width of a single NMR-line calculated from the relaxation time T_2 , as measured by the Carr-Purcell spin-echo technique (Sect. 3.7).

Because of the complicated structure of these spectra, we were not able to evaluate coupling constants by the least-square fitting routine.

3.6. Line Widths

To evaluate the relaxation time T_2 from the width of a NMR-line, a very homogeneous field B_0 is necessary, particularly if there are narrow NMR-lines.

By the rotation of the probe, field gradients perpendicular to the axis of rotation are eliminated; the line widths are only influenced by an inhomogeneity parallel to this axis. The ⁷Li-probe⁹, which is used to measure and to stabilize the field B_0 , can also be used to measure the inhomogeneity. By a good adjustment of the nickel-shims, a variation of B_0 in the range of the probe sample $\Delta B/B_0 \approx 10^{-7}$ can be achieved.

The influence on the line width of a constant field gradient parallel to the direction of the field B_0 has been calculated by DOLEGA¹⁸. According to this result, the measured line widths, which are not too much influenced by the inhomogeneity of B_0 , can be corrected to get the natural line widths, as they would

be recorded in an absolutely homogeneous field. In Table 2, the line widths are given with these corrections, and also the relaxation times T_2 calculated with the aid of Eq. (1).

Germanium methoxide $\text{Ge}(\text{OCH}_3)_4$ shows a significant dependence of the line width on the temperature Θ :

$$\Delta\nu_{1/2} = 22.2 \text{ Hz} - (0.16 \text{ Hz}/^\circ\text{C}) \cdot \Theta,$$

in the range $0 \leq \Theta \leq 50^\circ\text{C}$. The error is maximum $\pm 0.7 \text{ Hz}$.

In the range $20 \dots 40^\circ\text{C}$ the line widths of the germanium-tetra-alkyls did not show a dependence on the temperature within the limits of error.

Table 2. Line widths, relaxation times T_2 calculated from line widths and measured by the Carr-Purcell spin-echo method. The measurements were performed at a temperature $\Theta = 40^\circ\text{C}$ for GeI_4 and at $\Theta = 30^\circ\text{C}$ for the other compounds. The errors of the line widths and the related T_2 values are three times the r.m.s. error plus a systematical error of $\pm 0.04 \text{ Hz}$ due to the uncertainty of the inhomogeneity-correction and due to the fluctuations of B_0 in time. The errors of the Carr-Purcell results are the maximum errors.

Compound	Line Width $\Delta\nu_{1/2}/\text{Hz}$	T_2/msec from the line width	T_2/msec measured by the Carr- Purcell spin- echo-techn.
GeCl_4	2.02 ± 0.10	158 ± 8	163 ± 20
GeBr_4	1.76 ± 0.14	181 ± 15	196 ± 30
GeI_4^a	2.2 ± 0.3	145 ± 23	
$\text{GeI}_4^{a,b}$	32	10	
$\text{Ge}(\text{OCH}_3)_4$	17.7 ± 0.8^c	(18.0 ± 0.9)	
$\text{Ge}(\text{OCH}_3)_4$	12.2 ± 1.5^d	26 ± 3	30 ± 3
GeMe_4	0.58 ± 0.16	550 ± 200	740 ± 80
GeEt_4	15.6 ± 0.4^c	(20.4 ± 0.5)	140 ± 20
GePro_4	13.3 ± 0.2^c	(23.9 ± 0.4)	100 ± 12
GeBu_4	13.8 ± 0.5^c	(23.1 ± 0.8)	65 ± 7

^a The NMR-line widths of GeI_4 in CS_2 and benzene are the same.

^b Weaker line.

^c Width of the absorption curve of the unresolved spectrum.

^d Width of the single line calculated with the fitting routine.

3.7. Relaxation Times T_2

The relaxation times T_2 of germanium compounds were also determined by the spin-echo technique of CARR and PURCELL¹⁹. The inhomogeneity of the rf-field B_1 in the range of the probe sample, causes different precession angles during the rf-irradiation at different points of the sample. This effect, intensified by the B_0 -inhomogeneity, would cause a decrease of the echos after successive pulses even with-

¹⁸ U. DOLEGA, Ann. Phys. **16**, 153 [1955].

¹⁹ H. Y. CARR and E. M. PURCELL, Phys. Rev. **94**, 630 [1954].

out transverse relaxation. This decrease was calculated²⁰, and the measured values were corrected.

All measurements were performed at a temperature of the sample of 30 °C; the results are given in column 4 of Table 2; the errors are the maximum errors.

Comparing the T_2 values attained by the spin-echo technique of Carr and Purcell with those calculated from the line widths, there is agreement within the limits of error with GeCl_4 , GeBr_4 , and GeMe_4 . The

relaxation time T_2 of $\text{Ge}(\text{OCH}_3)_4$, measured by the Carr-Purcell technique, corresponds to that calculated for a single line by the least square fitting routine. The relaxation times of the other germanium alkyls do not correspond with the widths of the NMR-lines, as these lines consist in an unresolved spectrum of many single NMR-lines.

We like to thank Prof. Dr. H. KRÜGER for his support of this work, Dr. O. LUTZ for many helpful discussions, and the Deutsche Forschungsgemeinschaft for financial support.

²⁰ W. SAHM, Diplomarbeit, Tübingen 1970.

Estimating the Alpha-Particles Transmission through a Barrier

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Among the numerous doubts associated with the estimation of the alpha-decay penetrabilities, two queries have been selected in order to investigate their justification: is the ambiguous determination of α -nucleus potential a real obstacle and is the JWKB an inadequate approximation?

In the study of alpha decay, theoretical successes have been invariably achieved only in reproducing the general features of experimental data, i. e. in explaining behaviours but not absolute values. Thus the first great accomplishment was the explanation of the marked dependence of the halflife upon the energy of the emitted α -particle as due to the energy dependence of the penetrability factor^{1,2} and the second was the explanation of the reduced widths varying with the mass number as due to the variation of the α -particle formation probability with the level assignments of the most loosely bound four nucleons³.

As for absolute values, no approach has given a satisfactory explanation, and every time the authors, in their conclusions, nurtured the hope that the discrepancy recorded was largely due to some insufficiencies in the determination of the penetrabilities. The impossibility of unequivocal determination of the α -nucleus potential, the inaccuracy of the first-

order JWKB approximation, the nonlocality of the potential barrier, the uncertainties in assessing the decay energies, the neglect or approximate consideration of the coupling between different possible channels, are only some of the numerous insufficiencies invoked⁴⁻⁸.

The first two have been taken up in this letter in order to establish if these can really be held responsible for the said discrepancy, of some orders of magnitude, existing between the theoretical and the experimental absolute values of the decay constant. For this purpose we have applied the theoretical considerations briefly reviewed below to the ground-state transitions in even-even polonium isotopes.

Alpha-decay constant has long been considered⁹ to be a product of the probability of α -particle formation inside the nucleus and the probability of where R_{in} and R_{out} are the internal and outer turning points given by $Q(r) = 0$.

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¹ G. GAMOV, Z. Physik 51, 204 [1928].

² U. E. CONDON and R. W. GURNEY, Nature London 122, 439 [1928].

³ H. J. MANG, Phys. Rev. 119, 1069 [1960].

⁴ H. J. MANG, Ann. Rev. Nucl. Sci. 14, 1 [1964].

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⁶ GY. BENCZE, Physics Letters 23, 713 [1967].

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